

**Lewis Structures**  
**CHEMISTRY 110**

Name \_\_\_\_\_  
last first

Draw the electron dot structure for the following compounds. **You must show all electrons, bonding and nonbonding:**

Br <sub>2</sub>	PCl <sub>3</sub>
H <sub>2</sub> Se	SCl <sub>2</sub>
NO <sub>2</sub> F	H <sub>3</sub> PO <sub>3</sub>

HPPH (P <sub>2</sub> H <sub>2</sub> )	CN <sup>-</sup>
SO <sub>2</sub>	HBrO
IO <sub>2</sub> <sup>-</sup>	H <sub>2</sub> CO <sub>3</sub>
OOO (O <sub>3</sub> )	COCl <sub>2</sub>