em 100			Name _		
ie date	)			(last)	(first)
points		Lecture Section # Instructor		nstructor	
the family ( belongs. If	alkali metal, alkaline eart	h metal, trai s not have a	nsition metal common nai	, halogen, or nobleme, write <b>only</b> the	e <b>only</b> the common name e gas) to which the eleme group number (for examp ass.
SYMBOL	NAME OF ELEMENT (spelling counts!)	ATOMIC NUMBER	ATOMIC MASS	METAL OR NONMETAL	FAMILY (Group)
F					
Cu					
Si					
Ca					
Fe					
Ar					
Al					
Na					
S					
Au					
correspondi	ELEMENTS. Write the ng formulas which show DIATOMIC ELEMENT	the molecula Γ <u>NAME</u>	r compositio		diatomic elements and
4.					

2. Be, Ca, Ba

3. F, Br, I

4. B, Al, Ga