$\qquad$

1. How many moles are in $\mathbf{2 2 4}$ grams of carbon?
2. How many grams are in 4.87 moles of iron?

Answer $\qquad$

Answer $\qquad$
3. How many atoms are in $\mathbf{4 5}$ moles of silver?

Answer $\qquad$
4. How many grams are in 34 moles of water $\mathrm{H}_{2} \mathrm{O}$ ?

Answer $\qquad$
5. $\mathbf{5 4}$ moles of sucrose $\mathrm{C}_{12} \mathrm{H}_{22} \mathrm{O}_{11}$ is how many molecules?

Answer $\qquad$
6. $\mathbf{5 4}$ moles of KCl is how many grams?

Answer $\qquad$
7. $5.23 \times 10^{24}$ atoms of nitrogen is how many moles?

Answer $\qquad$
8. 5 moles of water has how many moles of hydrogen?

Answer $\qquad$
9. What is the mass in grams of 5.6 moles of $\mathbf{N a C l}$ ?

Answer $\qquad$
10. 28 grams of gold contains how many atoms of gold?

Answer
11. Give the name and formula of the compound that forms between the following elements

| Elements | Formula | Name |
| :--- | :--- | :--- |
| Potassium and oxygen | $\mathrm{K}_{2} \mathrm{~S}$ | Potassium Oxide |
| Sodium and sulfur |  |  |
| Calcium and oxygen |  |  |
| Aluminum and bromine |  |  |
| Lithium and nitrogen |  |  |
| Magnesium and iodine |  |  |
| Aluminum and phosphorous |  |  |

