PRACTICE PROBLEMS: NOMENCLATURE OF INORGANIC COMPOUNDS

1. $BaCO_3$
2. $FeSO_4$
3. S_2F_{10}
4. K_2O
5. HNO_2
6. $AlPO_4$
7. AuI_3
8. I_2O_5
9. Co_2O_3
10. $Cu(OH)_2$
11. NaF
12. NH_4HSO_3
13. NBr_3
14. $SnCl_4$
15. MgS
16. H_2CO_3
17. $Pb(ClO_3)_2$
18. Si_2F_6
19. $Ca(C_2H_3O_2)_2$
20. KOH
21. $Fe(NO_3)_3$
22. P_4S_7

1. barium carbonate
2. iron (II) sulfate
3. disulfur decafluoride
4. potassium oxide
5. hydrogen nitrite
6. aluminum phosphate
7. gold (III) iodide
8. diiodine pentoxide
9. cobalt (III) oxide
10. copper (II) hydroxide
11. sodium fluoride
12. ammonium bisulfite
13. nitrogen tribromide
14. tin (IV) chloride
15. magnesium sulfide
16. hydrogen carbonate
17. lead (II) chlorate
18. disilicon hexafluoride
19. calcium acetate
20. potassium hydroxide
21. iron (III) nitrate
22. tetraphosphorus heptasulfide